

# Ce-Fe-P (Cerium-Iron-Phosphorus)

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Recently, Chikhrii et al. [1997Chi] determined isothermal sections for this system at  $\sim 800$  °C for 0-30 at.% Ce and at  $\sim 590$  °C for 30-70 at.% Ce.

## Binary Systems

The Ce-Fe phase diagram [1993Zha] depicts two stoichiometric compounds,  $\text{Ce}_2\text{Fe}_{17}$  and  $\text{CeFe}_2$ , both forming peritectically.  $\text{Ce}_2\text{Fe}_{17}$  has two crystal modifications (hexagonal and rhombohedral).  $\text{CeFe}_2$  has the cubic  $\text{MgCu}_2$  type structure. The Ce-P phase diagram is not known. Three intermediate compounds are known: CeP (cubic, NaCl type),  $\text{CeP}_2$  (monoclinic), and  $\text{CeP}_3$  (monoclinic). A partial phase diagram is known for the Fe-P system [1982Kub]. The intermediate compound  $\text{Fe}_3\text{P}$  forms through a peritectic reaction at 1166 °C between liquid and  $\text{Fe}_2\text{P}$ .  $\text{Fe}_2\text{P}$  forms congruently at 1370 °C.  $\text{Fe}_3\text{P}$  is body-centered tetragonal with the  $\text{Ni}_3\text{P}$  type of structure.  $\text{Fe}_2\text{P}$  has the hexagonal C22 structure. The other intermediate phases at higher P contents are FeP (orthorhombic MnP type),  $\text{FeP}_2$  [orthorhombic  $\text{FeS}_2$  (marcasite) type], and  $\text{FeP}_4$  (monoclinic).

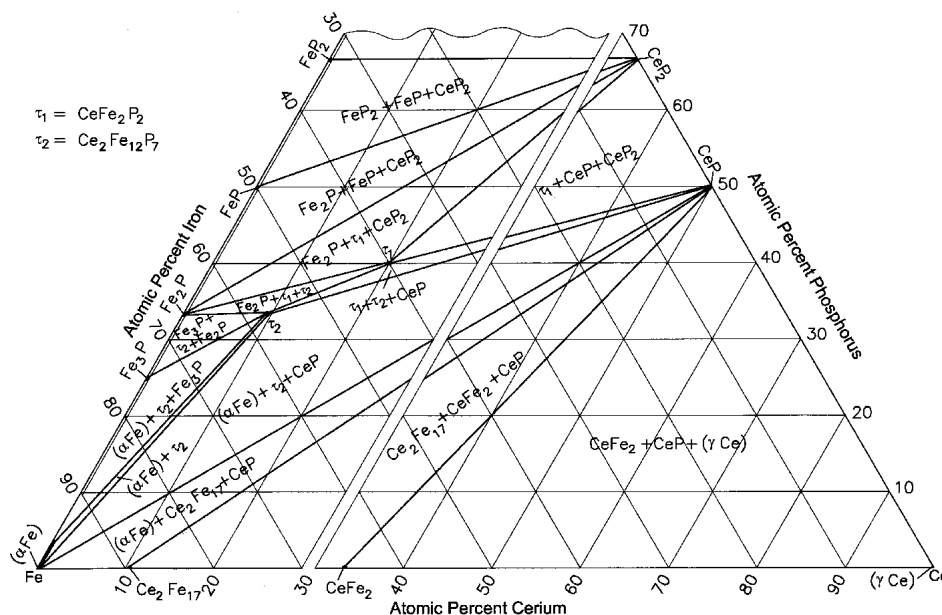
## Ternary Compounds

Three ternary compounds are known in this system.  $\text{CeFe}_2\text{P}_2$  ( $\tau_1$ ) has the  $\text{Al}_4\text{Ba}$ -type tetragonal structure [1985Jei].  $\text{Ce}_2\text{Fe}_{12}\text{P}_7$  ( $\tau_2$ ) has the  $\text{Zr}_2\text{Fe}_{12}\text{P}_7$ -type hexagonal

structure [1984Jei].  $\text{CeFe}_4\text{P}_{12}$  ( $\tau_3$ ) has the  $\text{LaFe}_4\text{P}_{12}$ -type cubic structure [1977Jei]. Table 1 lists the structural details of these compounds.

## Ternary Isothermal Section

With starting materials of purity of 99.9% Ce, 99.99% Fe, and 99.89% P, [1997Chi] prepared 36 alloy compositions. For compositions up to 33.3 at.% P, the alloys were melted in an arc furnace. For higher P contents, the powder mixtures were sintered by slow heating up to 800 °C. The samples were finally annealed for 500 h at 1070 K for compositions with  $\leq 30$  at.% Ce and at 870 K for compositions with 30-70 at.% Ce and quenched in water. The phase equilibria were studied by x-ray powder diffraction. The isothermal sections determined by [1997Chi] at 797 °C (1070 K) for 0-30 at.% Ce and at  $\sim 590$  °C for 30-70 at.% Ce are presented as a composite section in Fig. 1. The temperature of anneal for the higher Ce content alloys is assumed here to be just below the eutectic temperature (592 °C) of the Fe-Ce system, as no liquid phase is indicated by [1997Chi]. The ternary compounds  $\tau_1$  and  $\tau_2$  are present at 797 °C. The composition of the ternary compound  $\tau_3$  falls outside the range investigated by [1997Chi]. No homogeneity ranges were found for the ternary compounds. The third component solubility in the binary compounds is negligible.



**Fig. 1** Ce-Fe-P isothermal section at 797 °C (0-30 at.% Ce) and at  $\sim 590$  °C (30-70 at.% Ce) [1997Chi]. Narrow two-phase regions around tie-triangles are omitted.

**Table 1 Ce-Fe-P Crystal Structure and Lattice Parameter Data**

Phase	Composition, at. %	Pearson Symbol	Space Group	Prototype	Lattice Parameter, nm	Reference
CeFe <sub>2</sub> P <sub>2</sub> (τ <sub>1</sub> )	20 Ce 40 P	<i>I</i> 10	<i>I</i> 4/ <i>mmm</i>	Al <sub>4</sub> Ba	<i>a</i> = 0.3852 <i>c</i> = 1.0314	1985Jei
Ce <sub>2</sub> Fe <sub>12</sub> P <sub>7</sub> (τ <sub>2</sub> )	9.5 Ce 33.3 P	<i>hP</i> 21	<i>P</i> $\bar{6}$	Zr <sub>2</sub> Fe <sub>12</sub> P <sub>7</sub>	<i>a</i> = 0.9135 <i>c</i> = 0.3677	1984Jei
CeFe <sub>4</sub> P <sub>12</sub> (τ <sub>3</sub> )	5.9 Ce 70.6 P	<i>cI</i> 34	<i>Im</i> $\bar{3}$	LaFe <sub>4</sub> P <sub>12</sub>	<i>a</i> = 0.7792 ...	1977Jei

**References**

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